Data collection: XSCANS (Siemens, 1994). Cell refinement: XSCANS. Data reduction: XSCANS. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1985). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: ZORTEP (Zsolnai, 1997). Software used to prepare material for publication: PARST (Nardelli, 1983, 1995).

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3α -Hydroxy- 5α -androstane-4,17-dione

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Abstract

The asymmetric unit of the title compound, $C_{19}H_{28}O_3$, contains two independent molecules with almost identical geometry. The molecules have a planar 5α config-

© 1999 International Union of Crystallography Printed in Great Britain – all rights reserved uration as a result of a *trans-A/B* junction of the rings. The compound crystallizes in a triclinic cell, which is unusual for steroids. The molecules are linked head-to-tail *via* hydrogen bonds between the ring A hydroxyl group and the ketone group of ring D, forming two independent chains running along the c axis.

Comment

Our interest in preparing steroid aromatase inhibitors (Campos Neves et al., 1999; Tavares da Silva et al., 1996, 1997), clinically used as antitumor agents especially in the treatment of estrogen-positive breast cancers, led us to the title compound, 3α -hydroxy- 5α androstane-4,17-dione, (I). 3α , 4β -Dihydroxy- 5α -androstan-17-one, a key intermediate in the aforementioned synthesis, was selectively oxidized at C₄ with oxone, a mixed persulfate reagent, to give (I). The best conditions to perform this reaction (Tavares da Silva, 1997) use the oxidant in the presence of hydrated alumina (Hirano et al., 1991) under ultrasonic irradiation. Furthermore, an efficient and environmentally friendly chemical process has been achieved for this particular transformation, using the combination of a solid supported reagent and sonochemistry.



The unit cell contains two crystallographically independent molecules (1 and 2) with almost identical geometry. An ORTEPII (Johnson, 1976) view of the two molecules with the atom-numbering scheme is shown in Fig. 1. The internal degree of isostructurality between the two molecules as defined by Kálman et al. (1991) is given by I_D^{25} (distances) = 99.5% and I_D^{23} (valency angles) = 99.5%. The trans-A/B ring junctions produce almost planar molecules [bowing angle $6.43(10)^{\circ}$ for molecule 1 and $7.8(2)^{\circ}$ for molecule 2], with an angle of $19.02(6)^{\circ}$ between the average leastsquares planes of molecules 1 and 2. The distances between terminal O atoms were found to be 9.707(6) and 9.657(6) Å, and the values of the pseudo-torsion angles C19-C10-C13-C18 are 3.1 (2) and 1.7 (2)°, respectively, for molecules 1 and 2, indicating that molecule 1 is slightly more twisted than molecule 2. Rings A, B and C have slightly flattened chair conformations evidenced by the mean values of their torsion angles being less than 60% [57(2), 56.0(6) and 55(5)° for molecule 1, and 57 (2), 56.0 (8) and 55 (5)° for molecule 2]. Ring D adopts a less common 13β , 14α -half-chair conformation distorted towards a 14 α -envelope, as can be seen from

 $C_{19}H_{28}O_3$



Fig. 1. ORTEPII (Johnson, 1976) plot of the two symmetry independent molecules of the title compound. Displacement ellipsoids are drawn at the 50% probability level.

the asymmetry parameters ΔC_s (Duax & Norton, 1975) and pseudo-rotation parameters Δ and φ_m (Altona *et al.*, 1968): molecule 1, $\Delta C_s(14) = 10.6 (3)^\circ$, $\Delta C_2(13, 14) =$ 7.1 (4)°, $\varphi_m = 42.4 (3)^\circ$ and $\Delta = -12.2 (6)^\circ$; molecule 2, $\Delta C_s(14) = 10.5 (4)^\circ$, $\Delta C_2(13, 14) = 6.8 (4)^\circ$, $\varphi_m = 41.3 (3)^\circ$ and $\Delta = -12.1 (7)^\circ$.

The bond lengths and angles in (I) are within the normal ranges expected for this type of compound (Allen *et al.*, 1987). The molecules are linked head-to-tail *via* hydrogen bonds between the hydroxyl group of ring A and the ketone atom of ring D, forming chains running along the c axis. Each chain is built exclusively of molecules 1 or 2, and there are no strong hydrogen bonds between the two crystallographically independent molecules. The strength of the hydrogen bonds is similar in the two chains. Remarkably, the ketone O4 atoms are not involved in hydrogen bonding, except possibly for rather weak C—H···O interactions. Interactions between the chains appear to be mainly of the van der Waals type.

It should be stated that because none of the atoms is a strong enough anomalous scatterer at the Mo $K\alpha$ wavelength, the absolute configuration was not determined from the X-ray data, and the assumed chirality of the molecule is that determined from the route of synthesis.

Experimental

To a solution of 3α , 4β -dihydroxy- 5α -androstan-17-one (50 mg, 0.16 mmol) in dichloromethane containing a small amount of methanol (1.7 ml), wet alumina (335 mg) and oxone (252 mg, 0.41 mmol) were added. The suspension, under an inert atmosphere, was sonicated in a cleaning bath ('Bandelin', 450 W, 35 kHz) at 295 K until the steroid was consumed

(30 min, thin-layer chromatography). After dilution with dichloromethane, the suspension was filtered, washed with aqueous NaHCO3 and water, dried over magnesium sulfate and evaporated to dryness to give a crude material (43 mg, 87%) from which the title compound was isolated by crystallization. NMR: $\delta_{\rm H}$ (500 MHz, CDCl₃, Me₄Si) 0.77 (3H, s, 19-H₃), 0.87 (3H, s, 18-H₃), 2.09 (1H, ddd, $J_{16\alpha,16\beta} = 19.0$, $J_{16\alpha,15\beta} =$ 9.5, $J_{16\alpha,15\alpha}$ = 9.5 Hz, 16 α -H), 2.14 (1H, d, $J_{3\alpha-OH,3\beta-H}$ = 3.5 Hz, 3α -OH; disappears with D₂O), 2.45 (1H, dd, $J_{16\beta,16\alpha}$ = 19.0, $J_{16\beta,15\beta} = 9.0$ Hz, 16β -H), 2.86 (1H, dd, $J_{5\alpha,6\beta} = 12.5$, $J_{5\alpha,6\alpha} = 3.0 \text{ Hz}, 5\alpha \text{-H}, 4.0 (1\text{H}, ddd, J_{3\beta,2\beta} = 3.0, J_{3\beta,2\alpha} = 3.0, J_{3\beta,2\alpha} = 3.0, J_{3\beta,2\alpha} = 3.0 \text{ Hz}, 5\alpha \text{-H}, 4.0 (1\text{H}, ddd, J_{3\beta,2\beta} = 3.0, J_{3\beta,2\alpha} = 3.0 \text{ Hz}, 5\alpha \text{-H}, 5\alpha \text{-H},$ 3.0, $J_{3\beta-H,3\alpha-OH} = 3.5$ Hz, 3β -H; collapses with D₂O to a dd of $J_{3\beta,2\beta} = 3.0$ and $J_{3\beta,2\alpha} = 3.0$ Hz); $\delta_{\rm C}$ (50.3 MHz, CDCl₃, Me4Si) 13.9 (CH3, C19), 14.0 (CH3, C18), 20.0 (CH2), 21.1 (CH₂), 21.9 (CH₂), 29.4 (CH₂), 30.3 (CH₂), 31.7 (CH₂), 32.2 (CH₂), 34.8 (CH), 36.0 (CH₂), 43.2 (C_q), 47.9 (C_q), 51.4 (CH), 53.6 (CH), 54.6 (CH), 74.1 (CH, C3), 213.6 (Cq, C4), 221.3 (Cq, C17); m/z (EI): 304.2 (M⁺, 100%). Crystals of good quality suitable for X-ray analysis were grown from a mixture of ethyl acetate and hexane.

Crystal data

$C_{19}H_{28}O_3$	Mo $K\alpha$ radiation
$M_r = 304.43$	$\lambda = 0.71073 \text{ Å}$
Triclinic	Cell parameters from 25
<i>P</i> 1	reflections
a = 7.869 (2) Å	$\theta = 8.92 - 12.43^{\circ}$
b = 10.113(3) Å	$\mu = 0.081 \text{ mm}^{-1}$
c = 10.771(5) Å	T = 293 (2) K
$\alpha = 76.21 (3)^{\circ}$	Prism
$\beta = 85.78 (3)^{\circ}$	$0.51 \times 0.24 \times 0.12 \text{ mm}$
$\gamma = 84.60 (3)^{\circ}$	Colourless
$V = 827.6 (5) \text{ Å}^3$	
Z = 2	
$D_x = 1.222 \text{ Mg m}^{-3}$	
D_m not measured	
Data collection	
Enraf–Nonius CAD-4	$R_{\rm int} = 0.016$
diffractometer	$\theta_{\rm max} = 27.41^{\circ}$

Profile data from $\omega - 2\theta$ scans Absorption correction: none 4431 measured reflections 4377 independent reflections 2741 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $(\Delta/\sigma)_{max}$ $R[F^2 > 2\sigma(F^2)] = 0.039$ $\Delta\rho_{max} = 0$ $wR(F^2) = 0.113$ $\Delta\rho_{min} = -5$ S = 1.005Extinction4377 reflectionsScattering403 parametersInternalH atoms constrainedCrystal $w = 1/[\sigma^2(F_o^2) + (0.0577P)^2 + 0.0792P]$ where $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.196 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.133 \text{ e } \text{\AA}^{-3}$ Extinction correction: none Scattering factors from *International Tables for Crystallography* (Vol. C)

 $h = -10 \rightarrow 10$

 $k = -13 \rightarrow 12$

 $l = -13 \rightarrow 13$

3 standard reflections

frequency: 180 min

intensity decay: 1.3%

Table 1. Selected geometric parameters (Å, °)

C3-03 C4-04 C15-C16 C16-C17 C17-017	1.429 (4) 1.213 (4) 1.540 (5) 1.509 (5)	C3'-O3' C4'-O4' C15'-C16' C16'-C17'	1.415 (5) 1.209 (4) 1.532 (6) 1.501 (6)
C2-C3-C4 C5-C4-C3 C16-C17-C13	1.214 (4) 108.7 (3) 114.6 (3) 108.6 (3)	C17017 C4'C3'C2' C5'C4'C3' C16'C17'C13'	1.218 (4) 108.7 (3) 114.5 (3) 108.5 (3)

Table 2. Hydrogen-bonding geometry (Å, °)

D — $\mathbf{H} \cdot \cdot \cdot \mathbf{A}$	D—H	$\mathbf{H} \cdot \cdot \cdot \mathbf{A}$	$D \cdot \cdot \cdot A$	D — $H \cdot \cdot \cdot A$
O3—H28· · ·O17'	0.82	2.04	2.813 (4)	158
O3'—H28'···O17''	0.82	1.98	2.790 (4)	168
Symmetry code: (i) x .	x, 1 + z			

Examination of the structure with *PLATON* (Spek, 1998) showed that there were no solvent-accessible voids in the

crystal lattice. Data collection: CAD-4 Software (Enraf-Nonius, 1989). Cell refinement: CAD-4 Software. Data reduction: HEL-ENA (Spek, 1997). Program(s) used to solve structure: SHELXS97 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL97 (Sheldrick, 1997). Molecular graphics: OR-TEPII (Johnson, 1976). Software used to prepare material for publication: SHELXL97.

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Triterpenoide. XVII.† 3,28-Dimethoxy-3,11,28-trioxo-2,3-seco-18 β -olean-12-en-2,9 α -olid

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Abstract

The structure of the title compound, $C_{32}H_{46}O_7$, previously derived by spectroscopic methods, is confirmed. Steroidal ring *C* has a slightly distorted sofa form and all other six-membered rings have a chair conformation. Rings *B* and *C* are trans-fused, and rings *D* and *E* are cis-fused. The lactone ring has a conformation intermediate between envelope and half-chair.

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